IN THE CLAIMS

- 1. (original) A method of depositing a metallic film on a substrate comprising:
 - a. placing a substrate comprising an upper surface, a lower surface, and silicon in a reaction cell, wherein at least one of said surfaces is coated with a coating selected with from the group consisting of TaN, TiN, Ta, WN, WCN, TaSiN, and TiSiN;
 - b. injecting a source metal into the cell through the use of a carrier gas that is bubbled through water into the cell during a first pulse of 1-20 seconds duration;
 - c. injecting an inert gas into the cell during a second pulse of 1-10 seconds duration;
 - d. injecting a reducing agent into the cell during a third pulse of 1-10 seconds duration; and
 - e. injecting an inert gas into the cell during a fourth pulse of 1-10 seconds duration.
- 2. (original) The method of claim 1, wherein the reducing agent is selected from the group consisting of alcohols and aldehydes.
- 3. (original) The method of claim 1, wherein the source metal comprises a copper II compound.
- 4. (original) The method of claim 3, wherein the source metal is a hydrated copper II compound.
- 5. (original) The method of claim 1, wherein the carrier gas is an inert gas.
- 6. (original) The method of claim 1, wherein the carrier gas is argon.
- 7. (original) The method of claim 1, wherein the carrier gas is hydrogen.

- 8. (original) The method of claim 1, wherein the source metal comprises an anhydrous copper compound.
- 9. (original) The method of claim 1, wherein the source metal comprises a copper beta-diketonates.
- 10. (original) The method of claim 1, wherein the source metal comprises a silver I compound.
- 11. (original) The method of claim I, wherein the source metal comprises a silver II compound.
- 12. (original) The method of claim 1, wherein the source metal comprises a copper I compound.
- 13. (original) The method of claim 1, wherein said inert gas is selected from a group consisting of nitrogen, argon and helium.
- 14. (original) The method of claim 1 wherein said coating has a thickness in the range of 5-100 nanometers.
- 15. (canceled)
- 16. (canceled)
- 17. (canceled)
- 18. (canceled)

19. (canceled)

- 20. (original) A method of depositing a metallic film on a substrate comprising:
 - a. placing a substrate comprising an upper surface, a lower surface, and silicon in a reaction cell, wherein at least one of said surfaces is coated with a coating having a thickness in the range of 5-100 nanometers, and selected with from the group consisting of TaN, TiN, Ta, WN, WCN, TaSiN, and TiSiN;
 - b. injecting a source metal into the cell through the use of an inert carrier gas that is bubbled through water into the cell during a first pulse of 1-20 seconds duration;
 - c. injecting an inert gas into the cell during a second pulse of 1-10 seconds duration;
 - d. injecting a reducing agent selected from the group consisting of alcohols and aldehydes into the cell during a third pulse of 1-10 seconds duration;
 and
 - e. injecting an inert gas into the cell during a fourth pulse of 1-10 seconds duration.
- 21. (original) A method of depositing a metallic film on a substrate comprising:
 - a. placing a substrate comprising an upper surface, a lower surface, and silicon in a reaction cell, wherein at least one of said surfaces is coated with a coating selected with from the group consisting of TaN, TiN, Ta, WN, WCN, TaSiN, and TiSiN;
 - b. injecting a source metal into the cell through the use of a carrier gas that is bubbled through water into the cell during a first pulse;
 - c. purging excess source metal by injecting an inert gas into the cell during a second pulse;
 - d. injecting a reducing agent into the cell during a third pulse; and
 - e. removing excess reducing agent by injecting an inert gas into the cell during a fourth pulse.

- 22. (original) The method of claim 21, wherein the reducing agent is selected from the group consisting of alcohols and aldehydes.
- 23. (original) The method of claim 21, wherein the carrier gas is an inert gas.
- 24. (original) The method of claim 21, wherein said source metal is selected from a group consisting of a silver I compound, a silver II compound, a copper I compound, and a copper beta-diketonates.

Respectfully submitted,

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